Chapter 06

Spreadsheets (Calculating the Cost of the Base)

#### **6.1** The calculation scheme

### Available NaOH solutions from the XYZ company

Preparation	Concentration	Preparation	Concentration
A	5.0	K	30.0
В	7.5	L	32.5
C	10.0	M	35.0
D	12.5	N	37.5
E	15.0	0	40.0
F	17.5	P	42.5
G	20.0	Q	45.0
Н	22.5	R	47.5
I	25.0	S	50.0
J	27.5	T	52.5

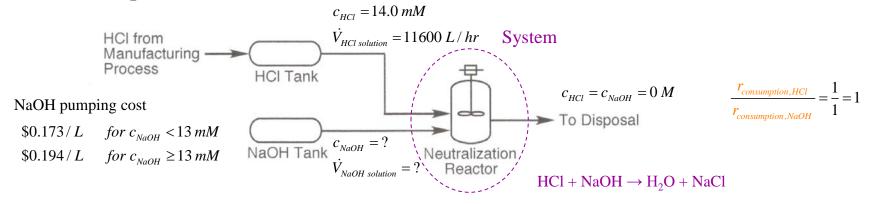
### Price formula depending on NaOH concentration

$$L = 0.058 (c_{NaOH})^{1.4} + 0.017$$

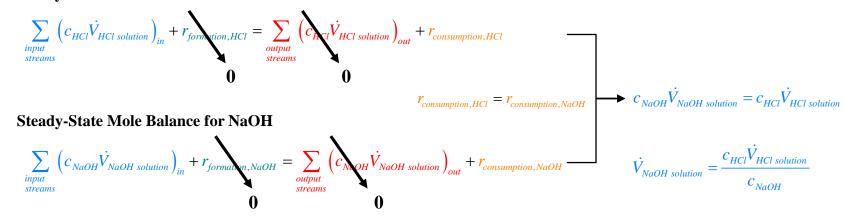
 $c_{NaOH}$ : concentration of NaOH in units of mM

#### **6.1** The calculation scheme

#### From chapter 5...



#### **Steady-State Mole Balance for HCl**



### **6.1** The calculation scheme

### Several possible strategies for calculations

- ➤ Hand-written calculation sheet
- ➤ Algebraic cost function
- > Computer program
- > Spreadsheet

## **6.2** Setting up a spreadsheet

#### Formatting and headings

- ➤ In cell C1, enter an overall title "NaOH Costs" and change it to Bold style.
- ➤ In cells A2 through F2, put the headings for the columns.
- ➤ In cells A3 through F3, enter the appropriate units.
- > Use the center-justify tool.
- > Adjust some column widths.
- ➤ Choose the font and its size (10-point Calibri)

		Α	В	C	D	E	F	
1	1			NaOH Cos	sts			
2	2	Conc.	Flow	Price	Cost	Pmp Cost	Tot Cost	
3	3	(mM)	(L/hr)	(\$/L)	(\$/hr)	(\$/hr)	(\$/hr)	
2	1							
5	5							
		I						

column	width	column	width
A	4	G	1
В	6	Н	12
C	6	I	5
D	7	J	3
E	7	K	1
F	7	L	4

### **6.2** Setting up a spreadsheet

#### 6.2.1 Setting up the spreadsheet for the acid-neutralization problem

- ➤ NaOH concentrations (Column A):
  - 1. In cell A5, enter the lowest concentration (5.0).
  - 2. Set the number format to display one digit to the right of the decimal point.
  - 3. Select A6 and type the formula "=A5+2.5".
  - 4. Drag the handle from A6 through A24.
  - 5. Put the given information.

	_					_						
	Α	В	C	D	E	F	G	Н	1	J	K	L
1			NaOH Co	sts								
2	Conc.	Flow	Price	Cost	Pmp Cost	Tot Cost		HCl flow (L/hr) =	11600			
3	(mM)	(L/hr)	(\$/L)	(\$/hr)	(\$/hr)	(\$/hr)		HCl conc (mM) =	14			
4												
5	5.0							NaOH cost (\$/L) =				
6	7.5							0.0058	*C^	1.4	+	0.017
7	10.0											
8	12.5							Pumping cost (\$/L	) =			
9	15.0							C<13 mM	0.173			
10	17.5							C>=13 mM	0.194			
11	20.0											

(6.2)

#### ➤ NaOH Flow rate (Column B):

- 1. Select B5 and type the formula "=I3\*I2/A5" (see equation 6.2).
- 2. Drag the handle from B5 through B24.
- 3. Select B5 and type a dollar sign (\$) in front of the number "=I\$3\*I\$2/A5"
- 4. Drag the handle from B5 through B24 again.
- 5. Set comma style and remove decimal points.

### **6.2** Setting up a spreadsheet

#### 6.2.1 Setting up the spreadsheet for the acid-neutralization problem

- ➤ NaOH Price (Column C):
  - 1. Select C5 and type the formula "=H\$6\*A5^J\$6+L\$6" (see equation 6.1).  $\frac{1}{L} = 0.058(c_{NaOH})^{1.4} + 0.017$
  - 2. Drag the handle from C5 through C24.
  - 3. Set the number format to display three digits to the right of the decimal point.
  - 4. Set dollar sign (\$) in front of the number.

#### **➤ NaOH Purchase Cost (Column D):**

- 1. Select D5 and type the formula "=B5\*C5".
- 2. Drag the handle from D5 through D24.
- 3. Set comma style and remove decimal points.
- 4. Set dollar sign (\$) in front of the number.

### **6.2** Setting up a spreadsheet

#### 6.2.1 Setting up the spreadsheet for the acid-neutralization problem

- **▶** NaOH Pumping Cost (Column E):
  - 1. Select E5 and type the formula "=IF(A5<13,I\$9\*B5,I\$10\*5)".
  - 2. Drag the handle from E5 through E24.
  - 3. Set comma style and remove decimal points.
  - 4. Set dollar sign (\$) in front of the number.
- > Total Cost (Column F):
  - 1. Select F5 and type the formula "=D5+E5".

Given NaOH pumping cost

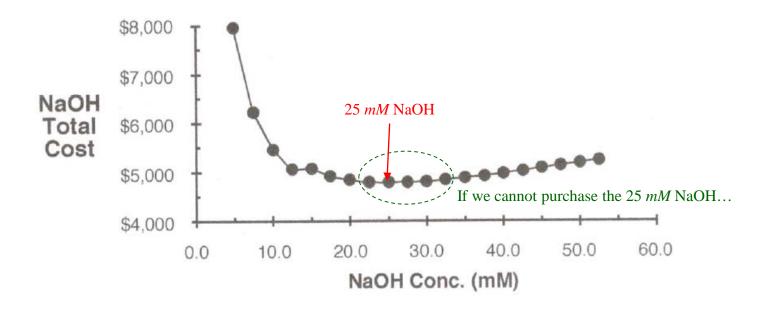
\$0.173/L for  $c_{NaOH} < 13 \, mM$ 

 $$0.194/L \quad for c_{NaOH} \ge 13 \, mM$ 

	Α	В	С	D	Е	F	G	Н	1	J	K	L
1			NaOH Costs									
2	Conc.	Flow	Price	Cost	Pmp Cost	Tot Cost		HCl flow (L/hr) =	11600			
3	(mM)	(L/hr)	(\$/L)	(\$/hr)	(\$/hr)	(\$/hr)		HCl conc (mM) =	14			
4								_				
5	5.0	32,480	\$0.072	\$2,345	\$5,619	\$7,964		NaOH cost (\$/L) =			_	,
6	7.5	21,653	\$0.114	\$2,477	\$3,746	\$6,223		0.0058	*C^	1.4	+	0.017
7	10.0	16,240	\$0.163	\$2,642	\$2,810	\$5,452						
8	12.5	12,992	\$0.216	\$2,808	\$2,248	\$5,055		Pumping cost (\$/L	)=			
9	15.0	10,827	\$0.274	\$2,967	\$2,100	\$5,067		C<13 mM	0.173			
10	17.5	9,280	\$0.336	\$3,117	\$1,800	\$4,918		C>=13 mM	0.194			
11	20.0	8,120	\$0.401	\$3,260	\$1,575	\$4,835		•				
12	22.5	7,218	\$0.470	\$3,395	\$1,400	\$4,796		Ch	0000	٥h	1.	
13	25.0	6,496	\$0.542	\$3,524	\$1,260	\$4,784		CII	ange	at	110	5
14	27.5	5,905	\$0.617	\$3,646	\$1,146	\$4,792						
15	30.0	5,413	\$0.695	\$3,764	\$1,050	\$4,814						
16	32.5	4,997	\$0.776	\$3,876	\$969	\$4,845						
17	35.0	4,640	\$0.859	\$3,984	\$900	\$4,884						
18	37.5	4,331	\$0.944	\$4,088	\$840	\$4,928						
19	40.0	4,060	\$1.032	\$4,188	\$788	\$4,976						
20	42.5	3,821	\$1.122	\$4,286	\$741	\$5,027						
21	45.0	3,609	\$1.214	\$4,380	\$700	\$5,080						
22	47.5	3,419	\$1.308	\$4,471	\$663	\$5,134						
23	50.0	3,248	\$1.404	\$4,559	\$630	\$5,189						
24	52.5	3,093	\$1.502	\$4,645	\$600	\$5,245						

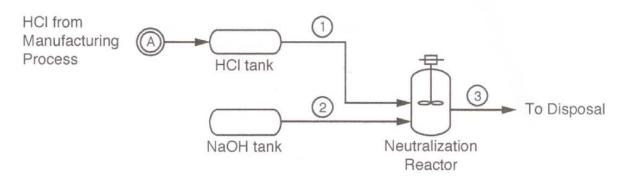
## 6.3 Graphing

#### Chart of total cost versus NaOH concentration



## 6.3 Graphing

### Process flow diagram with stream table



Flows kg/h				
Line no. Stream Component	1 Acid feed	2 Base feed	3 Reactor outlet	ABC Chemical Co.
HCl	6	_	_	Acid neutralization
NaOH	_	6	-	1x10 <sup>8</sup> L/yr
H <sub>2</sub> O	11594	6490	18096	Sheet no. 1
Total	11600	6496	18096	Dwg by Date
				Checked 1 Sep.2010